STEREOCHEMISTRY OF LARDOLURE THE AGGREGATION PHEROMONE OF THE ACARID MITE, LARDOGLYPHUS KONOI[†]

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Abstract -- (1R,3R,5R,7R)-Stereochemistry was assigned to lardolure (1,3,5,7-tetramethyldecyl formate) on the basis of GLC comparison of the natural pheromone with the synthetic pheromones prepared in partially stereocontrolled manners.

Lardolure was isolated by Y. Kuwahara et al. as the aggregation pheromone of the acarid mite Lardoglyphus konoi (Sasa et Asanuma) (Acarina: Acaridae), a primary pest for stored products with high protein content such as dried meat and fish meal. Its structure was determined spectroscopically as 1a and confirmed by synthesizing a diastereomeric mixture of $1a.^1$ In addition, (R)-configuration was assigned to C-1 of 1a by the comparison of the natural pheromone with (S)-1-methylheptyl formate. Their optical rotations as well as the NMR spectra measured in the presence of a chiral shift reagent were compared. However, the relative stereochemistry of the pheromone 1a remained unknown. In connection with our synthetic studies on optically active pheromones, we became interested in establishing the stereochemistry of this pheromone.

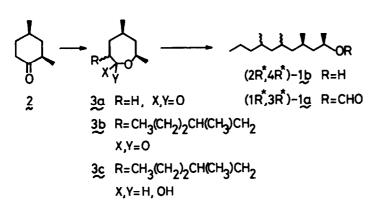
Because of the presence of four chiral centers in 1a, eight diastereomers possessing (\underline{R})-configuration at C-1 are possible for the structure of lardolure. Therefore we must determine the relative configuration of lardolure. The key to this problem was found in Y. Kuwahara's observations: (1) the synthesized mixture consisting of eight diastereomers of 1a gave seven peaks when analyzed by capillary GLC; (2) among them, the peak having the shortest retention time coincided with that of the natural pheromone. By using these interesing observations, we planned to determine the stereochemical relationship among the chiral centers.

Our strategy is based on the following consideration: if a mixture consisting of 1,3-syn-diastereomers of 1a can be prepared stereoselectively and the natural pheromone coincides in GLC retention time with one of them, it is possible to assign 1,3-syn-relationship to the natural pheromone. The same consideration is applicable to both C-3,5-

[†]Pheromone Synthesis Part 92. Part 91, K. Mori and T. Ebata, <u>Tetrahedron</u> in press. The experimental part of this work was taken from a part of the forthcoming doctral thesis of S. K. (1996).

relationship and C-5,7-relationship, so that it must be feasible to correlate the configuration of the four Me substituents of 1a. In order to obtain 1,3-<u>syn</u>-dimethyl building blocks required for the preparation of 1,3-<u>syn</u>-1a, 3,5-<u>syn</u>-1a and 5,7-<u>syn</u>-1a, catalytic hydrogenation of polymethylphenols was chosen by virtue of its <u>cis</u>-selectivity and simplicity.

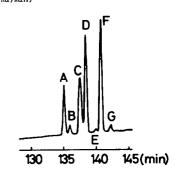
Our synthesis of 1,3-syn-lardolure (1 R^* ,3 R^*)-1a is shown in Scheme 1. Catalytic hydrogenation of 2,4-dimethylphenol was followed by Jones' oxidation to give the known ketone 2 (cis:trans=85:15). On oxidation with MCPBA, this ketone gave 3a in 94 % yield. Alkylation of 3a with 1-iodo-2-methylpentane afforded 3b in 51 % yield. This was reduced with DIBAL-H to give a mixture of a lactol 3c and the corresponding hydroxy aldehyde. The Wolff-Kishner reduction of the mixture gave ($2R^*$,4 R^*)-1b, which was esterified with HCO2H to give 1,3-syn-lardolure (1 R^* ,3 R^*)-1a in 63 % yield from 3b. The result of its GLC analysis is shown in Fig. 1, where four large peaks and three small peaks can be seen clearly in agreement with Y. Kuwahara's observation. Since this synthesis started from the cis-rich ketone 2 (85:15), the four large peaks A, C, D and F, should be ascribed to 1,3-syn-1a and the three small peaks, B, E and G, to 1,3-syn-1a. Among them, peak A showed the same retention time as the peak of the natural pheromone by the coinjection analysis of (1 R^* ,3 R^*)-1a with the natural pheromone. 1,3-syn-Relationship was therefore assigned to the natural pheromone.



Scheme 1. Synthesis of 1,3-syn-lardolure.

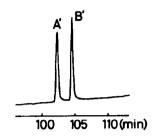
isomers of 1,3-<u>syn</u>-lardolure. (Column, OV-101, 50 m x 0.25 mm at 100° + 0.5°/min; Carrier gas, N₂, 28 ml/min)

Fig. 1. GLC separation of the



Next we turned our attention to the elucidation of the stereochemical relationship between C-3 and C-5 of 1a. Since the relative configuration between C-1 and C-3 was already assigned to be <u>syn</u>, we decided to synthesize $1,3,5-\underline{\text{syn}}$ -lardolure $(1\underline{R}^*,3\underline{R}^*,5\underline{R}^*)-1a$ instead of $3,5-\underline{\text{syn}}$ -lardolure (Scheme 2). Catalytic hydrogenation of mesitol gave a diastereomeric mixture of 2,4,6-trimethyl-1-cyclohexanol⁴ containing 4a (50 %) and 4b (44 %). The mixture was purified by SiO_2 chromatography to give crystalline 4a (35 % yield, mp. $68\sim70^\circ$) and oily 4b (39 % yield). On oxidation with Jones' reagent, both 4a and 4b yielded 4c as a single product in 89 % yield. In the 13 C NMR spectra of 4a, 4b and 4c, only six peaks were observed reflecting their symmetrical structures. Baeyer-Villiger oxidation of 4c gave 5 in 98 % yield. LAH reduction of 5 afforded a diol 6a and its primary OH group was tosylated selectively to give 6b in 66 % yield from 5. Coupling of 6b with 1-methylbutylmagnesium bromide⁵ followed by esterification gave $1,3,5-\underline{\text{syn}}$ -lardolure $(1\underline{R}^*,3\underline{R}^*,5\underline{R}^*)-1a$ as a mixture of epimers at C-7 in 41 % yield from 6b. As shown in Fig. 2, its gas chromatogram gave only two peaks (A' and B') as expected, and peak A' was identical in retention time with peak A shown in Fig. 1 by the coinjection analysis of

Pig. 2. GLC separation of the isomers of 1,3,5-syn-lardolure. (Column, OV-101, 50 m x 0.25 mm at 110° + 0.5°/min; Carrier gas, N₂, 27 ml/min)



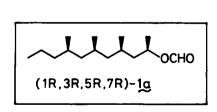
Scheme 2. Synthesis of 1,3,5-syn-lardolure.

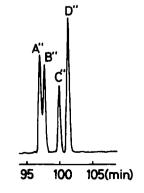
 $(1\underline{R}^*,3\underline{R}^*,5\underline{R}^*)$ -1a with $(1\underline{R}^*,3\underline{R}^*)$ -1a. Thus peak A' appeared at the same retention time as that of the natural pheromone, because peak A coincided in retention time with the peak of the natural pheromone. Since peak A' was due to one of the C-7 epimers of $(1\underline{R}^*,3\underline{R}^*,5\underline{R}^*)$ -1a, we reasoned that the natural pheromone also had $(1\underline{R}^*,3\underline{R}^*,5\underline{R}^*)$ -stereochemistry, i.e., 1,3,5-syn-relationship.

Fig. 3. GLC separation of the isomers of 5,7-syn-lardolure.

(Column, OV-101, 50 m x 0.25 mm at $120^{\circ} + 0.5^{\circ}/\text{min}$; Carrier gas, N_2 , 45 ml/min)

Scheme 3. Synthesis of 5,7-syn-lardolure.





Finally, we prepared 5,7-syn-lardolure (5 \underline{R}^* ,7 \underline{R}^*)-1a to clarify the relative stereochemistry between C-5 and C-7 of 1a (Scheme 3). Methanolysis of the lactone 5 gave 7a. This hydroxy ester 7a was treated successively with p-TsCl, LAH, p-TsCl and NaI to give 8c in 76 % yield from 5 via 7b, 8a and 8b. Alkylation of 4-pentanolide with the iodide 8c gave a mixture of cis- and trans-9a in 50 % yield (cis:trans=2:3). Reduction of this mixture with DIBAL-H was followed by the Wolff-Kishner reduction and esterification to give 5,7-syn-lardolure (5 \underline{R}^* ,7 \underline{R}^*)-1a as a mixture of four diastereomers in 61 % yield from 9a. As shown in Fig. 3, its gas chromatogram gave four peaks, A", B", C" and D". The peak A" appeared at the same retention time as the peak A shown in Fig. 1. by the coinjection analysis of (5 \underline{R}^* ,7 \underline{R}^*)-1a with (1 \underline{R}^* ,3 \underline{R}^*)-1a. From the similar consideration as described for (1 \underline{R}^* ,3 \underline{R}^* ,5 \underline{R}^*)-1a, the natural pheromone was proved to have 5,7-syn-stereochemical relationship.

On the basis of the three results described above, it was concluded that lardolure had 1,3,5,7- $\underline{\text{syn}}$ -relative stereochemistry. Furthermore, if we accept Y. Kuwahara's proposal concerning the ($\underline{\text{R}}$)-configuration at C-1, 1 lardolure must be (1 $\underline{\text{R}}$,3 $\underline{\text{R}}$,5 $\underline{\text{R}}$,7 $\underline{\text{R}}$)-1,3,5,7-tetramethyldecyl formate (1 $\underline{\text{R}}$,3 $\underline{\text{R}}$,5 $\underline{\text{R}}$,7 $\underline{\text{R}}$)-1a.

From a biosynthetic point of view, lardolure is presumed to belong to polyketide-derived natural products. As a similar polyketide-derived natural product, 2,4,6,8-tetramethylundecanoic acid is known as the major component of the preen gland waxes of the greylag $goose^{6,7}$. It is interesting to note that this acid also has (R)-configuration at every chiral center, i.e., (2R,4R,6R,8R)-stereochemistry.^{8,9}

In conclusion, we propose $(1\underline{R},3\underline{R},5\underline{R},7\underline{R})$ -stereochemistry for lardolure. In the following paper, the synthetic confirmation of this structure assignment for lardolure will be reported.

EXPERIMENTAL

All baps and maps were uncorrected. IR spectra were measured as films for oils or as nujol mulls for solids on a Jaco IRA-102 spectrometer. ¹H NNR spectra were recorded with TMS as an internal standard at 60 MHz in CCl₄ on a Hitachi R-24A spectrometer. ¹³C NMR spectra were recorded at 25 MHz in CDCl₃ with TMS as an internal standard on a JEOL JNM FX-100 spectrometer. Optical rotations were measured on a Jaco DIP 140 polarimeter. GLC analyses were performed on a Yanaco G-180 gas chromatograph. Merck Kieselgel 60 Art 7734 was used for SiO₂ column chromatography unless otherwise stated.

(4R*,6R*)-4-Methyl-6-heptanolide 3a. To a stirred mixture of 80 % MCPBA (3.9 g) and NaHCO₃ (1.9 g) in dry CH₂Cl₂ was added dropwise a soln of 2 (1.80 g) in dry CH₂Cl₂ (4 ml) at room temp. After stirring for 2.5 h, the precipitate was filtered off and the filtrate was washed with 10 % NaHSO₃ ag, sat NaHCO₃ ag and brine, dried (MgSO₄) and concentrated in vacuo. The residue was distilled to give 1.90 g (94 %) of 3a, bp. 94-96°/4 Torr; $n_1^{\rm O}$ 1.4525; wmax 2960 (s), 2930 (s), 2870 (s), 1730 (vs), 1450 (s), 1190 (s), 1160 (s), 1090 (s) cm⁻¹; δ 0.92 (3H, d, J=6 Hz), 1.26 (3H, d, J=6 Hz), 1.3-2.1 (5H, m) 2.1-2.4 (2H, m), 4.0-4.7 (1H, m); GLC (Column, OV-101, 50 m x 0.25 mm at 110° + 0.5°/min; Carrier gas, N₂, 43 ml/min): Rt 31.8 min (83 %); trans isomer at 33.3 min (17 %). (Pound: C, 67.15; H, 9.82. Calc for C₀H₁₄O₂: C, 67.57; H, 9.93 %). (2R*5*,4R*,6R*)-4-Methyl-2-((2R*5*)-2-methylpentyl]-6-heptanolide 3b. A soln of LDA was prepared by the addition of a soln of n-BuLi (1.75 N in n-hexane, 27.6 ml) to a stirred and cooled soln of (i-Pr)₂NH (5.16 g) in dry THF (45 ml) at 0° under Ar. HMPA (48 ml) was added to the soln at -5--10° and the mixture was cooled to -78°. To the resulting slurry was added dropwise a soln of 3a (6,00 g) in dry THF (45 ml) over a period of 75 min. After stirring for 25 min, the mixture was warmed to -45°. To the soln was added a soln of 1-iodo-2-methylpentane (11 g) in dry THF (25 ml) at -45--40°. The soln was allowed to stand for 2 days at -40°. It was then poured into ice-sat NH₄Cl aq and extracted with ether. The ether soln was washed with water and brine, dried (MgSO₄) and concentrated in vacuo. The residue (11.01 g) was chromatographed over SiO₂ (500 g, n-hexane-EtOAc) and distilled to give 4.88 g (51 %) of 3b, bp. 93-95°/0.15 Torr; n_0^{23} 1.4554; wmax 2960 (s), 2930 (s), 2870 (s), 1730 (s), 1460 (m), 1380 (m), 1185 (s), 1115 (s) cm⁻¹; δ 0.89 (3H, t, J=6 Hz), 0.92 (2x3H, d, J=6

 $C_{14}H_{26}O_{2}$: C, 74.28; H, 11.58 %).

(2R*,4R*,6R*5*,8R*5*)-4,6,8-Trimethyl-2-undecanol (2R*,4R*)-1b. A soln of DIBAL-H in n-hexane [12 ml of 25 (w/v)% soln) was added to a stirred and cooled soln of 3b (4.50 g) in dry toluene (130 ml) at -70° under Ar. After stirring for 30 min, sat NH₄Cl aq (10 ml) was added at -70°. The mixture was poured into ice-3N HCl and stirred for 30 min at room temp. It was extracted with ether. The ether soln was washed with sat NaHCO₃ aq and brine, dried (MgSO₄) and concentrated in vacuo to give 5.16 g of a mixture of 3c and a hydroxy aldehyde, vmax 3400 (m), 2960 (s), 2930 (s), 2870 (m), 2700 (w), 1725 (m) cm⁻¹; δ 0.7~1.1 (9H, m), 1.1~2.4 (16H, m), 3.39 (1H, s, OH), 3.5~4.3 (1H, m), 4.47 (0.67H, d, J=7 Hz), 9.43 (0.33H, s, br). This was dissolved in diethylene glycol (24 ml). 80 % N₂H₄·H₂O (6 g) was added to the soln and the mixture was heated under reflux for 1 h. KOH aq (6.0 g in 6 ml of water) was then added to the soln and the mixture was heated under reflux for 1.5 h. The excess N₂H₄·H₂O was distilled off and the mixture was stirred and heated for 1.5 h at 200°. After cooling, it was diluted with water and extracted with ether. The ether soln was washed with water and brine, dried (MgSO₄) and concentrated in vacuo. The residue (3.71 g) was chromatographed over SiO₂ (120 g; n-hexane-EtOAc) and distilled to give 3.22 g (76 % from 3b) of (2R*,4R*)-1b, b.p. 108~109°/1± Torr; n_0^{2} 1.4390; vmax 3350 (m), 2960 (s), 2930 (s),

Hz), 1.28 (3H, d, J=6 Hz), 1.1~2.1 (12H, m), 2.2~2.8 (1H, m), 4.1~4.7 (1H, m). (Found: C, 74.42; H, 11.56. Calc for

2870 (g), 1460 (m), 1380 (m), 1150 (m), 1115 (m), 1055 (m) cm^{-1} ; 6 0,7~1.05 (12H, m), 1.12 (3H, d, J=6 Hz), 1.05~2.0 (13H, m), 2.48 (1H, s, OH), 3.35~4.00 (1H, m). (Pound: C, 78.26; H, 14.26. Calc for $\text{C}_{14}\text{H}_{30}\text{O}$: C, 78.43; H, 14.11 %).

(1R*,3R*,5R*S*,7R*S*)-1,3,5,7-Tetramethyldecyl formate (1R*,3R*)-1a. A mixture of (2R*,4R*)-1b (1,50 g) and HOO2H (>98 % purity, 20 ml) was stirred for 75 min at 65°. It was then poured into ice-sat NaHCO3 and extracted with ether. The ether soln was washed with sat NaHCO3 and brine, dried (MgSO4) and concentrated in vacuo. The residue was chromatographed (Merck Lobar® column; n-hexane-Et₂O) and distilled to give 1,40 g (83 %) of (1R*,3R*)-1a, bp. 86-87°/0,9 Torr; ngl 1,4299; vmax 2970 (s), 2930 (s), 2880 (m), 2850 (m), 1730 (s), 1460 (m), 1380 (m), 1185 (s), 1130 (m) cm⁻¹; 6 (CDCl₃) 0,7-1,1 (12H, m), 1,25 (3H, d, J=6 Hz), 11-2,0 (13H, m), 4,95-5,45 (1H, m), 8,00 (1H, s); GLC (Column, OV-101, 50 m x 0,25 mm at 100° + 0,5°/min; Carrier gas, N₂, 28 ml/min): Rt 135,1 min (14 %), 135,9 min (3 %), 137,5 min (18 %), 138,3 min (30 %), 139,8 min (1 %), 140,7 min (32 %), 142,1 min (2 %). (Found: C, 74,33; H, 12,57. Calc for C₁₅H_{3O}O₂: C, 74,32; H, 12,48 %). The peak due to the natural pheromone coincided with that of the shortest Rt (135,1 min).

(11°,2R°,4r°,6S°)-2,4,6-Trimethyl-1-cyclohexanol 4a and its C-1 epimer 4b. A mixture of mesitol (10.0 g) and W-7 Raney Ni (2 g) in EtOH (3 ml) was stirred under a hydrogen pressure of 110 atm at 135° for 3 h. The catalyst was filtered off and the filtrate was concentrated in vacuo. The residue was diluted with ether. The ether soln was washed with 5 % NaOH aq, water and brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed over SiO₂ (Merck Kieselgel 60, Art 9385, 400 g). Elution with n-hexane-EtOAc (20:1) gave an oil, which was distilled to give 4,07 g (39 %) of 4b, b.p. $101\sim102^{\circ}/50$ Torr; $10^{\circ}/50$ 14531; vmax 3520 (m), 3000 (s), 2950 (s), 2910 (s), 1460 (m), 1380 (m), 1180 (m), 970 (s), 930 (m) cm⁻¹; 6 (CDCl₃) 0.87 (3H, d, J=5 Hz), 0.91 (2x3H, d, J=6 Hz), 1.1~1.8 (7H, m), 1.61 (1H, s, OH), 3.39 (1H, s, Wh/2=5 Hz); $1^{3}/50$ NMR & 18.57, 22.42, 32.41, 36.43, 37.26, 74.42. (Found: C, 76.18; H, 12.65. Calc for $C_{9}H_{18}$ 0: C, 75.99; H, 12.76 %). Further elution with n-hexane-EtOAc (15:1) gave a crystalline solid, which was recrystallized from n-hexane to give 3.63 g (35 %) of 4a as needles, m.p. 68~70°, vmax 3400 (m), 3000 (s), 2960 (s), 1460 (m), 1055 (m), 1045 (m) cm⁻¹; 6 (CDCl₃) 0.83 (3H, d, J=6 Hz), 0.95 (2x3H, d, J=5 Hz), 1.1~1.8 (7H, m), 2.31 (1H, s, OH), 2.57 (1H, t, J=8 Hz); $1^{3}/50$ NMR & 18.81, 22.05, 31.75, 39.43, 43.03, 82.00. (Found: C, 76.26; H, 12.72. Calc for $C_{9}H_{18}$ 0: C, 75.99; H, 12.76 %). GLC analysis of the crude reduction product (Column, OV-101, 50 m x 0.25 mm at 90°; Carrier gas, N₂, 45 ml/min): Rt 31.7 min (50 %, 4a), 32.4 min (44 %, 4b), 38.0 min (2 %), 39.2 min (4 %).

(2R*,4r*,6S*)-2,4,6-Trimethylcyclohexanone 4c. Jones reagent (30 ml) was added to a stirred and cooled soln of 4a (10.0 g) and 4b (7.0 g) in acetone (200 ml) at 0-5°. After stirring for 10 min, MeOH was added and the stirring was continued for 10 min. The mixture was extracted with ether. The ether soln was washed with water and brine, dried (MgSO₄) and concentrated in vacvo. The residue was distilled to give 15.0 g (89 %) of 4c, b.p. 77-80°/19 Torr; n_0^{19} 1.4423; vmax 2980 (s), 2890 (s), 1715 (s), 1460 (s), 1380 (m), 1130 (s), 1000 (m) cm⁻¹; δ (CDCl₃) 0.94 (3H, d, J=6 Hz), 0.96 (2x3H, d, J=6 Hz), 1.15~2.7 (7H, m); 13 C NMR δ 14.51, 21.21, 32.00, 44.20, 45.40, 214.30. (Found: C, 76.98; H, 11.76. Calc for CgH₁₆O: C, 77.09; H, 11.50 %).

 $\frac{(2R^*,4S^*,6S^*)-2,4-\text{Dimethyl}-6-\text{heptanolide}}{(40\text{ ml})}$ 5. To a stirred mixture of 85 % MCPBA (5.8 g) and NaHCO₃ (2.9 g) in dry CH₂Cl₂ (40 ml) was added dropwise a soln of 4c (3.20 g) in dry CH₂Cl₂ (7 ml) at room temp. After stirring for 4 h, the precipitate was filtered off and the filtrate was washed with 10 % NaHSO₃ aq, sat NaHCO₃ aq and brine, dried (MgSO₄) and concentrated in vacuo. The residue was distilled to give 3.50 g (98 %) of 5, b.p. 83-846/2.5 Torr; n₁09 1.4515; vmax 2990 (jm), 2940 (m), 1730 (s), 1455 (m), 1380 (m), 1200 (s), 1130 (m), 1095 (m), 1035 (m) cm⁻¹; 6 0.92 (3H, d, J=6 Hz), 1.10 (3H, d, J=7 Hz), 1.30 (3H, d, J=7 Hz), 1.3~2.2 (5H, m), 2.3~3.0 (1H, m), 4.2~4.75 (1H, m) (Found: C, 69.17; H, 10.30. Calc for C₉H₁₆O₂: C, 69.19; H, 10.32 %).

 $(2R^*,4S^*,6S^*)-2,4$ -Dimethyl-1,6-heptanediol 6a. A soln of 5 (9.00 g) in dry ether (20 ml) was added to a stirred and ice-cooled suspension of LAH (3.0 g) in dry ether (150 ml). The mixture was stirred for 1.5 h at room temp. The usual alkaline work-up gave an oil, which was distilled to give 8,93 g (97 %) of 6a, b.p. $101-102^\circ/0.95$ Torr; 10^{17} 1.4546; wmax 3350 (s), 2970 (s), 2880 (s), 1460 (m), 1375 (m), 1115 (m), 1055 (s), 1040 (s) cm⁻¹; 6 0.87 (2x3H, d, J=6 Hz), 1.21 (3H, d, J=6 Hz), 1.3-2.0 (6H, m), 3.30 (2H, d, J=5 Hz), 3.65 (2H, s, OH), 3.5-4.0 (1H, m). (Found: C, 67.11; H, 12.42. Calc for C_{91} -0.02; C, 67.45; H, 12.58 %).

 $\frac{(2\text{Re}^*,4\text{Re}^*,6\text{S}^*)-4,6-\text{Dimethyl}-7-\text{tosyloxy}-2-\text{heptanol}}{(7.00\text{ g})}$ in dry C₅H₅N (50 ml). After stirring for 3 h at 0-5°, the mixture was poured into ice-3N HCl and extracted with ether. The ether soln was washed with dil HCl, sat NaHCO₃ aq and brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed over SiO₂ (500 g, n-hexane-EtOAc) to give 9,30 g (68 %) of **6b**, vmax 3600(w), 3450 (w), 3000 (m), 2960 (m), 1610 (w), 1505 (w), 1465 (m), 1360 (s), 1190 (s), 1180 (s), 1100 (m), 965 (s) cm⁻¹. This was employed in the next step without further purification.

 $\frac{(2R^*,4R^*,6R^*,8R^*S^*)-4,6,8-\text{Trimethyl}-2-\text{undecanol}}{(7.41\text{ g})} \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*,6R^*,6R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard reagent was prepared from 2-bromopentane} \\ \frac{(2R^*,4R^*,6R^*)-1b.}{(2R^*,4R^*,6R^*)-1b.} \quad \text{A Grignard rea$

<u>Methyl</u> $(2R^*,4S^*,6S^*)$ -6-hydroxy-2,4-dimethylheptanoate 7a. To a soln of 5 (11.84 g) in abs MeCH (150 m1) was added conc H_2SO_4 (1.0g). The mixture was stirred for 1 h at room temp. After neutralization with sat NaHCO₃ ag, the mixture was concentrated in vacuo. The residue was diluted with water and extracted with ether. The ether soln was washed with water and brine, dried (MgSO₄) and concentrated in vacuo to give 15.09 g of crude 7a, vmax 3450 (m), 2980 (s), 2940 (s), 1730 (s), 1460 (m), 1190 (s), 1170 (s), 1140(s) cm⁻¹. This was employed in the next step without further purification.

<u>Methyl</u> $(2R^*,4S^*,6S^*)-2,4$ -Dimethyl-6-tosyloxyheptanoate **7b.** p-TsCl (30 g) was added to a stirred and ice-cooled soln of crude **7a** (15.09 g) in dry C_5H_5N (60 ml). After stirring for 4 h at $0-5^\circ$, the mixture was poured into ice-water and extracted with ether. The ether soln was washed with sat CuSO₄ aq, water and brine, dried (MgSO₄) and concentrated in vacuo to give 26.79 g of crude **7b.** vmax 1740 (s), 1360 (s), 1190 (s), 1170 (s), 900 (s) cm⁻¹. This was employed in the next step without further purification.

 $(2R^*,3R^*)-2,4$ -Dimethylheptyl tosylate 8b. p-TsCl (21 g) was added to a stirred and ice-cooled soln of 8a (8.0 g) in dry C₅H₅N (55 ml). After stirring for 3 h at 0-5°, the mixture was poured into ice-3N HCl and extracted with ether. The ether soln was washed with dil HCl, sat NaHCO₃ aq and brine, dried (MgSO₄) and concentrated in vacuo to give 19.0 g of crude 8b, was 2960 (s), 2930 (s), 2880 (m), 1595 (m), 1360 (s), 1190 (s), 1170 (s), 960 (s) cm⁻¹. This was employed in the next step without further purification.

(28*,48*)-1-Iodo-2,4-dimethylheptane 8c. NaI (19 g) was added to a soln of crude 8b (19.0 g) in dry acetone (250 ml). The mixture was stirred for 4 h under reflux and for 3 h at room temp. It was poured into ice-water and extracted with n-hexane. The n-hexane soln was washed with 10 % Na₂S2₂O₃ aq and brine, dried (MgSO₄) and concentrated in <u>vacuo</u>. The residue was chromatographed over SiO₂ (300 g, n-hexane) to give 13.1 g (93 % from 8a) of 8c, vmax 2960 (s), 2920 (s), 2870 (s), 1450 (m), 1375 (m), 1190 (m) cm⁻¹; δ 0.7~1.1 (9H, m), 1.1~1.7 (8H, m), 2.85~3.3 (2H, m). This was employed in the next step without further purification.

(2R*s*,4R*s*)-2-[(2R*,4R*)-2,4-dimethylheptyl]-4-pentanolide 9a. A soln of LDA was prepared by the addition of a soln of n-Buli (1.53 N in n-hexane, 30 ml) to a stirred and cooled soln of $(\underline{i}$ -Pr)₂NH (4.85 g) in dry THF (60 ml) at 0° under Ar. HMPA (40 ml) was added to the soln at -20° and the mixture was cooled to -69°. To the resulting slurry was added dropwise at -69° a soln of 4-pentanolide (4.0 g) in dry THF (40 ml) over a period of 2.5 h. After stirring for 15 min at -69°, the mixture was warmed to -45°. To the soln was added a soln of 8c (10.16 g) in dry THF (40 ml) at -40~45°. The soln was allowed to stand for 33 h at -40°. It was then poured into ice-sat NH₄Cl aq and extracted with ether. The ether soln was washed with water and brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed over SiO₂ (Merck Kieselgel 60 Art 9385, 450 g, n-hexane-EtOAc) and distilled to give 4.52 g (50 %) of 9a, b.p. 107~109°/0.15 Torr; n₀¹⁷ 1.4496; vmax 2980 (s), 2940 (s), 2880 (m), 1770 (s), 1460 (m), 1380 (m), 1180 (s), 1120 (m) cm⁻¹; δ 0.7~1.0 (9H, m), 1.0~2.1 (12H, m), 1.30 (1.35H, d, J=6 Hz), 1.34 (1.65H, d, J=6 Hz), 2.1~2.7 (1H, m), 4.0~4.7 (1H, m); GLC (Column, OV-101, 50 m x 0.25 mm at 120° + 0.5°/min; Carrier gas, N₂, 43 ml/min): Rt 90.1 min (45 %), 91.2 min (55 %). (Found: C, 74.12; H, 11.31. Calc for Cl₁H₂₆O₂: C, 74.28; H, 11.58 %).

(2R*s*,3R*s*,5R*s*)-2-hydroxy-5-methyl-3-[(2R*,4R*)-2,4-dimethylheptyl]tetrahydrofuran 9b. A soln of DIBAL-H in n-hexane [9.7 ml of 25 (w/v) % soln] was added to a stirred and cooled soln of 9a (3.50 g) in dry toluene (98 ml) below -63° under Ar. After stirring for 35 min at -65-70°, sat NH₄Cl was added to the mixture. It was poured into ice-3N HCl and stirred for 20 min at room temp. The mixture was extracted with ether. The ether soln was washed with sat NaHCO₃ aq and brine, dried (MgSO₄) and concentrated in vacuo to give 3.61 g of crude 9b, vmax 3400 (m), 2960 (s), 2920 (s), 1450 (m), 1370 (m), 995 (m) cm⁻¹. This was employed in the next step without further purification.

 $\frac{(2R^*S^*,4R^*S^*,6R^*,6R^*,6R^*)-4,6,3-\text{Trimethyl-2-undecanol}}{(3.61\ g)} \quad (6R^*,6R^*)-1b. \quad 80\ \&\ N_2H_4\cdot H_2O\ (4.1\ g) \quad \text{was added to a soln of crude 9b}$ (3.61 g) in diethylene glycol (16.5 ml). The mixture was heated under reflux for 1 h. KOH aq (4.1 g in 4.1 ml of water) was then added to the soln and it was heated under reflux for 1.5 h. The excess $N_2H_4\cdot H_2O$ was distilled off and the mixture was stirred and heated for 1.5 h at 200°. After cooling, the mixtur was diluted with water and extracted with ether. The ether soln was washed with water and brine, dried (MgSO₄) and concentrated in vacuo. The residue was chromatographed over SiO_2 (Merck Kieselgel 60 Art 9385, 130 g) n-hexane-EtOAc) and distilled to give 2.33 g (70 % from 9a) of $(6R^*,8R^*)-1b$, b.p. $105^{\circ}-106^{\circ}/2.4$ Torr; n_0^{21} 1,4390; vmax 3350 (m), 2970 (s), 2930 (s), 1460 (m), 1380 (m), 1150 (m), 1115 (m), 1055 (m) cm⁻¹; 6 0.7~1.0 (12H, m), 1.10 (3H, d), J=6 Hz), 1.0~1.9 (13H, m), 2.30 (1H, s, OH), 3.4~4.0 (1H, m); (Found: C, 78.23; H, 13.63. Calc for $C_{14}H_{30}O$: C, 78.43; H, 14.11 %).

 $\frac{(1R^*S^*, 3R^*S^*, 7R^*) - 1_4, 3_5, 7 - \text{Tetramethyldecyl}}{(1R^*S^*, 3R^*) - 1_6, (6R^*, 8R^*) - 1$

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